



April 2018, PSU-Phuket.

Workshop on How to Publish Papers in International Journals

Coordinator : Dr. Raymond J. Ritchie



How to make a poster for a conference.



Growth of Photosynthetic Bacteria (*Rhodospseudomonas palustris*) on Cooking Oils

Navata PHONGJARUS, Chanita SUVAPHAT, Naiyana SRICHAJ and Raymond J. RITCHIE*

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Abstract

Large volumes of food waste heavily loaded with cooking oil and liquid waste cooking oil are produced by restaurants, hotels and street food vendors. Much of the oil finds its way to waste treatment plants and drainages where oil and grease are environmentally damaging and costly to deal with. Photoheterotrophic growth of *Rhodospseudomonas palustris* (photosynthetic bacteria) (CGA 009) was used to decompose different types of cooking oil (palm and soybean oils; cooked and uncooked palm oil) under anaerobic and microaerobic conditions. Photosynthesis can be easily measured in *Rhodospseudomonas* using PAM technology (Pulse Amplitude Modulation Fluorometry). Optimum irradiance $\approx 284 \mu\text{mol photon m}^{-2} \text{s}^{-1}$ and maximum photosynthetic electron transport rate $341 \mu\text{mol e}^{-} \text{g}^{-1} \text{BChl a s}^{-1}$. *Rhodospseudomonas* used cooking oils as the sole carbon sources grown anaerobically in PM media with ammonia as the nitrogen source. *Rhodospseudomonas* grew equally well on fresh unused and on used palm oil as well as soybean oil and so used cooking oil is not toxic. Digestion of palm oil and soybean oil was much greater if 0.5% ethanol was added. The small amount of ethanol added apparently formed oil/ethanol micelles in the cell suspension which accelerated digestion of the palm oil and soybean oil. *Rhodospseudomonas* can produce small volumes of hydrogen gas from biodegradation of oil but only under nitrogen fixing conditions (anaerobic; no ammonia) but N-fixing cells grew slowly. A two-stage anaerobic digestion using yeast as the first stage, followed by a *R. palustris* digestion was tested. The yeast-step removed much of the reducing sugar and proteins and partially broke down some of the cooking oil. The second-stage *Rhodospseudomonas* photoferrmentation grew well taking advantage of the ethanol produced by the yeast but produced very little or no H_2 because of its organic nitrogen content.

Keywords: *Rhodospseudomonas*; Photosynthetic Bacteria; Cooking Oil; Biodegradation; Photosynthesis using PAM Fluorimetry.

Measuring Photosynthesis using a PAM machine

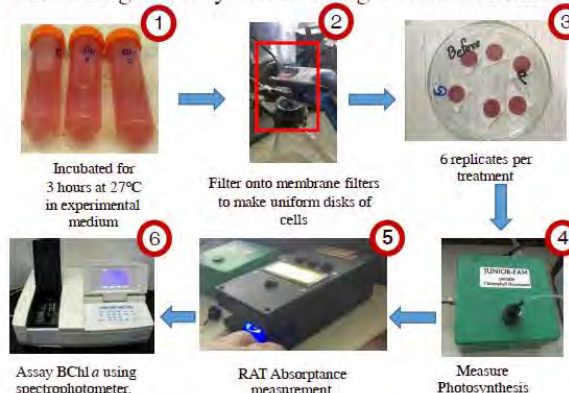


Fig. 1 Measuring photosynthesis of the photosynthetic bacterium using a Pulse Amplitude Modulation (PAM) Fluorometer

1. The cultures grown in PM medium (5ml) were filtered onto 6 membrane filters to produce a uniform disk of cells covering a known surface area. Sets of 6 disks were prepared for each experimental treatment. Disks kept in petri dishes with moist filter paper to prevent desiccation. 2. Measure Photosynthetic Electron Transport Rate (ETR) of the photosynthetic bacterium using PAM fluorimetry (Ritchie and Runcie, 2013; Ritchie and Mekinda, 2015). 3. RAT machine in blue light mode is used to measure absorbance (Ritchie and Runcie 2014). 4. Measure Bacteriochlorophyll spectrophotometrically to quote ETR on BChl a basis.

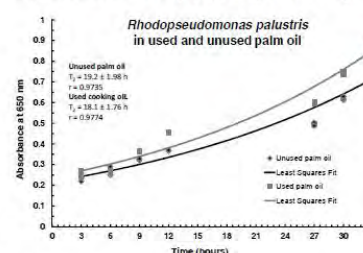


Fig. 2 Comparison of growth curves of *Rhodospseudomonas palustris* cells grown on unused fresh Palm Oil (10 ppt) and used Palm Oil (10 ppt) that had been used to fry chicken. Growth was followed as $A_{650 \text{ nm}}$ VS. time (h) using a spectrophotometer. *Rhodospseudomonas* will grow equally well on unused palm oil and oil that has been used for frying food ($t_2 \approx 18$ to 19 hours).

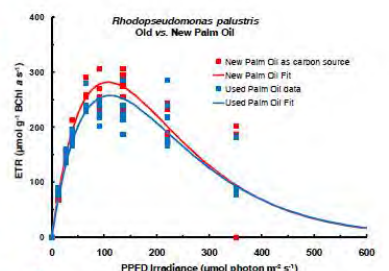


Fig. 3 Photosynthetic electron transport rate of *Rhodospseudomonas palustris* incubated on unused and used palm oil as a carbon source for 3 hours in PM medium. Cells had been grown in modified PM medium with unused Palm Oil as the sole carbon source.

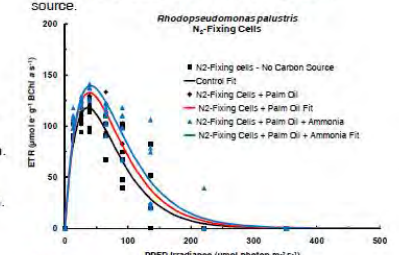


Fig. 4 shows that N-fixing cells have a lower photosynthetic rate than non-N fixing cells but were able to use Palm oil as a carbon source but did not respond quickly to added ammonia.

Highlights

- Liquid waste cooking oil is a major environmental problem.
- *Rhodospseudomonas* metabolises fresh and used Palm oil.
- *Rhodospseudomonas* can also digest Soy Bean Oil.
- Cooking oils can be broken down anaerobically or microaerobically.

References

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(2) Ritchie RJ and Runcie JW (2014) A portable Reflectance-Absorbance-Transmittance (RAT) meter for vascular Plant leaves. *Photosynthetica* 52: 614-626. DOI: 10.1007/s11099-014-0069-9
(3) Ritchie RJ, Mekinda N (2015). Measurement of photosynthesis using PAM technology in a purple sulphur bacterium *Thermochromatium tepidum* (Chromatiaceae). *Photochem. Photobiol.* 91: 350-358. DOI: 10.1111/php.12413



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Arsenic toxicity in a Photosynthetic Bacterial Symbiont of *Wolffia arrhiza*.

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Summary

An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia arrhiza* plants has been cultured and identified as a *Rhodospseudomonas* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a *Rhodospseudomonas*. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorbance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as $\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$. ETR vs. Irradiance (*E*) curves fitted the waiting-in-line model ($\text{ETR} = (\text{ETR}_{\text{max}} \times E/E_{\text{opt}}) \times \exp(1 - E/E_{\text{opt}})$). Yield (*Y*) was only ≈ 0.3 to 0.4 . *Rhodobacter* saturates at about 250 to 350 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ or $\approx 15\%$ sunlight and shows photoinhibition at high irradiances (overall E_{opt} was $298 \pm 7.36 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 642 \pm 10.6 \mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$; Alpha (α) = $6.05 \pm 0.200 \text{ e}^- \text{photon}^{-1} \text{m}^2 \text{g}^{-1} \text{BChl a}$). Photosynthetic performance was much worse in Low-P medium ($n = 108$, overall $E_{\text{opt}} = 158 \pm 15.4 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 194 \pm 13.5 \mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$; $\alpha = 3.30 \pm 0.400 \text{ e}^- \text{photon}^{-1} \text{m}^2 \text{g}^{-1} \text{BChl a}$). *Rhodobacter* is resistant to As(V) toxicity up to at least 1 mol m^{-3} in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was $497 \pm 100 \text{ mmol m}^{-3}$ but there is a threshold effect below about 200 mmol m^{-3} As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

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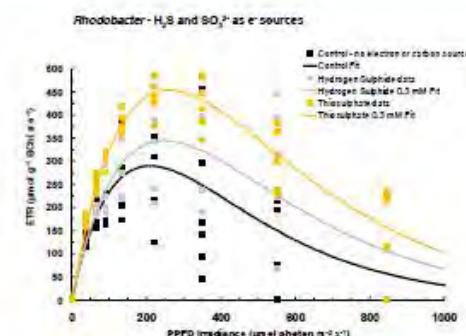


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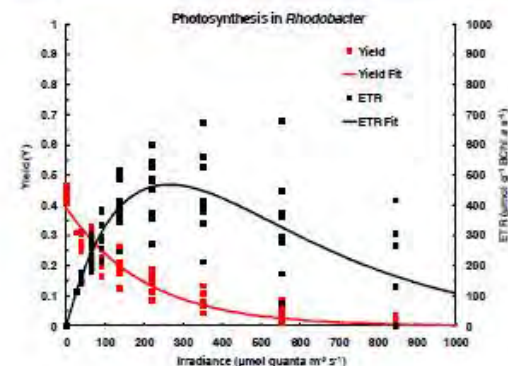


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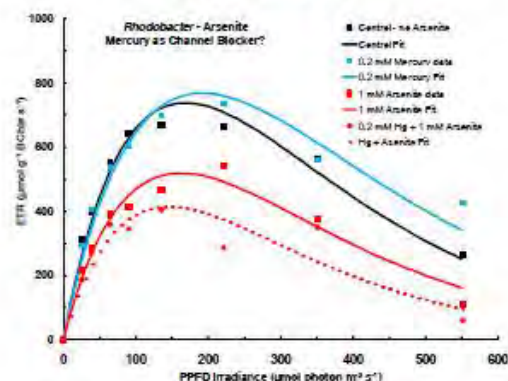


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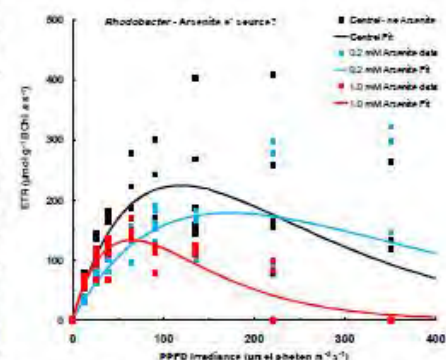


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Conclusions

- Arsenite (As(III)) is toxic but Arsenate (As(V)) has no short-term inhibitory effects. Response to As(V) is different under high P compared to low-P conditions.
- $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.
- As(III) does not act as an electron source for photosynthesis in *Rhodobacter*.

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- (1) Ritchie RJ, Ruxton JW (2013). Measurement of the Photosynthetic Electron Transport Rate in an Anoxygenic Photosynthetic Bacterium *Azolla* (*Rhodospseudomonas*) marina using PAM Fluorescence. *Photosynth. Photochem.* 99: 370-383.
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- (4) Ritchie, Raym and F. and Meljinda, N (2016). Arsenic Toxicity in the Water Weed *Wolffia arrhiza* Measured using Pulse Amplitude Modulation Fluorescence (PAM) Measurements of Photosynthesis. *Ecotoxicology and Environmental Safety* 132: 178 – 187; doi:10.1016/j.ecoenv.2016.06.004
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Must have university label



Try and find a picture of you that does not make you look like a idiot. I find that very hard. My colleague is luckier.

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Underline who is presenting the poster.

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More Bits and Pieces

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An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia aarhiza* plants has been cultured and identified as a *Rhodopseudomonad* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a *Rhodopseudomonad*. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorptance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as $\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$. ETR vs. Irradiance (*E*) curves fitted the waiting-in-line model ($\text{ETR} = (\text{ETR}_{\text{max}} \times E/E_{\text{opt}}) \times \exp(1 - E/E_{\text{opt}})$). Yield (*Y*) was only ≈ 0.3 to 0.4 . *Rhodobacter* saturates at about 250 to 350 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ or $\approx 15\%$ sunlight and shows photoinhibition at high irradiances (overall E_{opt} was $298 \pm 7.36 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 642 \pm 10.6 \mu\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$; $\text{Alpha}\alpha$ (α) = $6.05 \pm 0.200 \text{ e}^- \text{ photon}^{-1} \text{ m}^2 \text{g}^{-1} \text{BChl } a$). Photosynthetic performance was much worse in Low-P medium ($n = 108$, overall $E_{\text{opt}} 158 \pm 15.4 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 194 \pm 13.5 \mu\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$; $\alpha = 3.30 \pm 0.400 \text{ e}^- \text{ photon}^{-1} \text{ m}^2 \text{g}^{-1} \text{BChl } a$). *Rhodobacter* is resistant to As (V) toxicity up to at least 1 mol m^{-3} in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was $497 \pm 100 \text{ mmol m}^{-3}$ but there is a threshold effect below about 200 mmol m^{-3} As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

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Pretty Pictures

Wolffia arrhiza

Flower
Stamen
& Anthers

Daughter
frond

Frond

Adaxial
Surface
with
Stoma

~ 1 mm

Daughter
frond

Developing Seeds

Wettable Abaxial Surface

This mm ruler gives you some idea of its size.
You can do experiments on a real flowering plant on a
very small scale, even in a single Eppendorf Tube (1.4
ml).
Generation time \approx 2-5 days.

Bad news:
Despite what you would think
Wolffia is not easy to grow!

Water meal
Wolffia columbiana
Photo by Vic Ramey
© 2000 University of Florida



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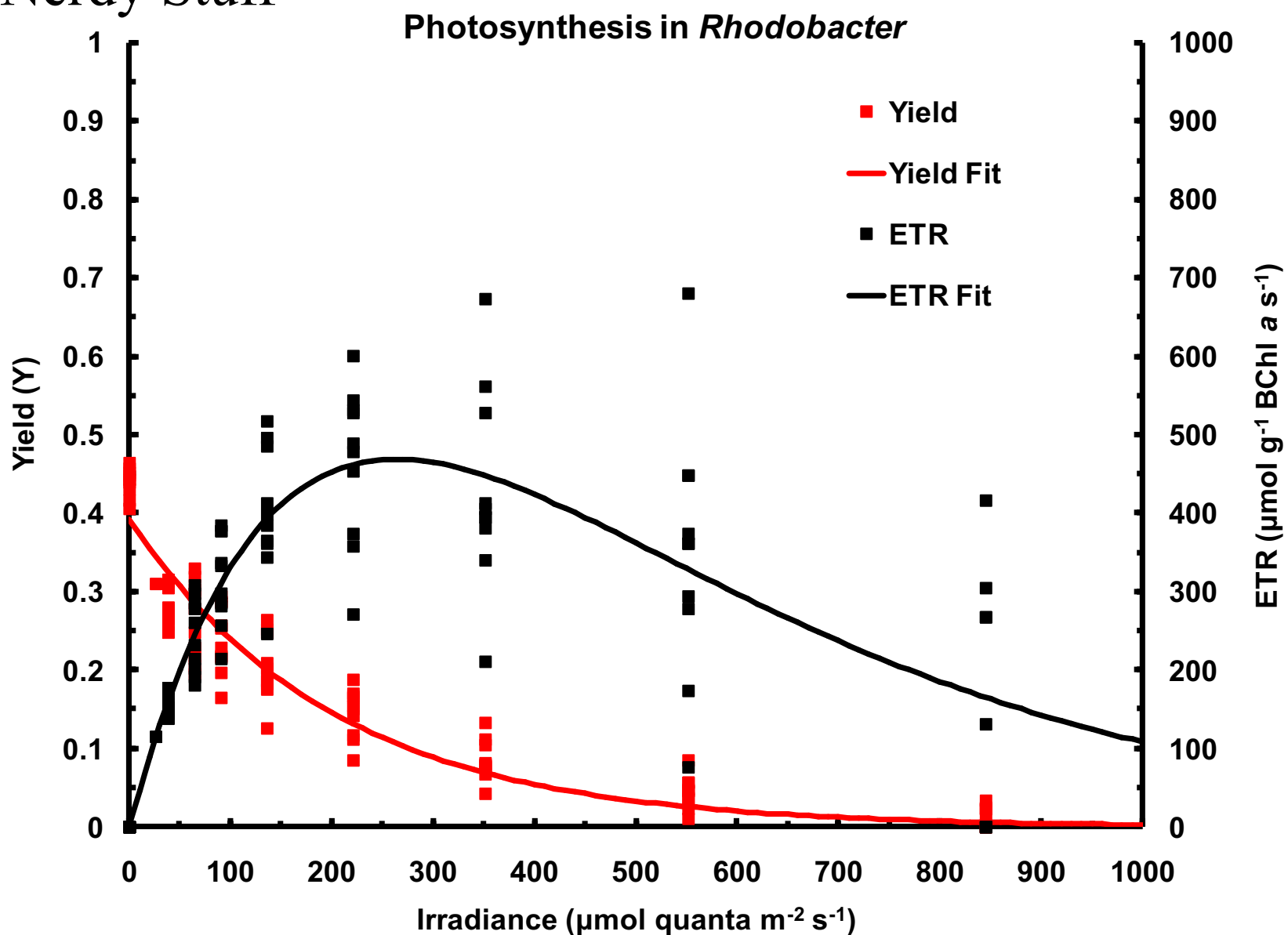


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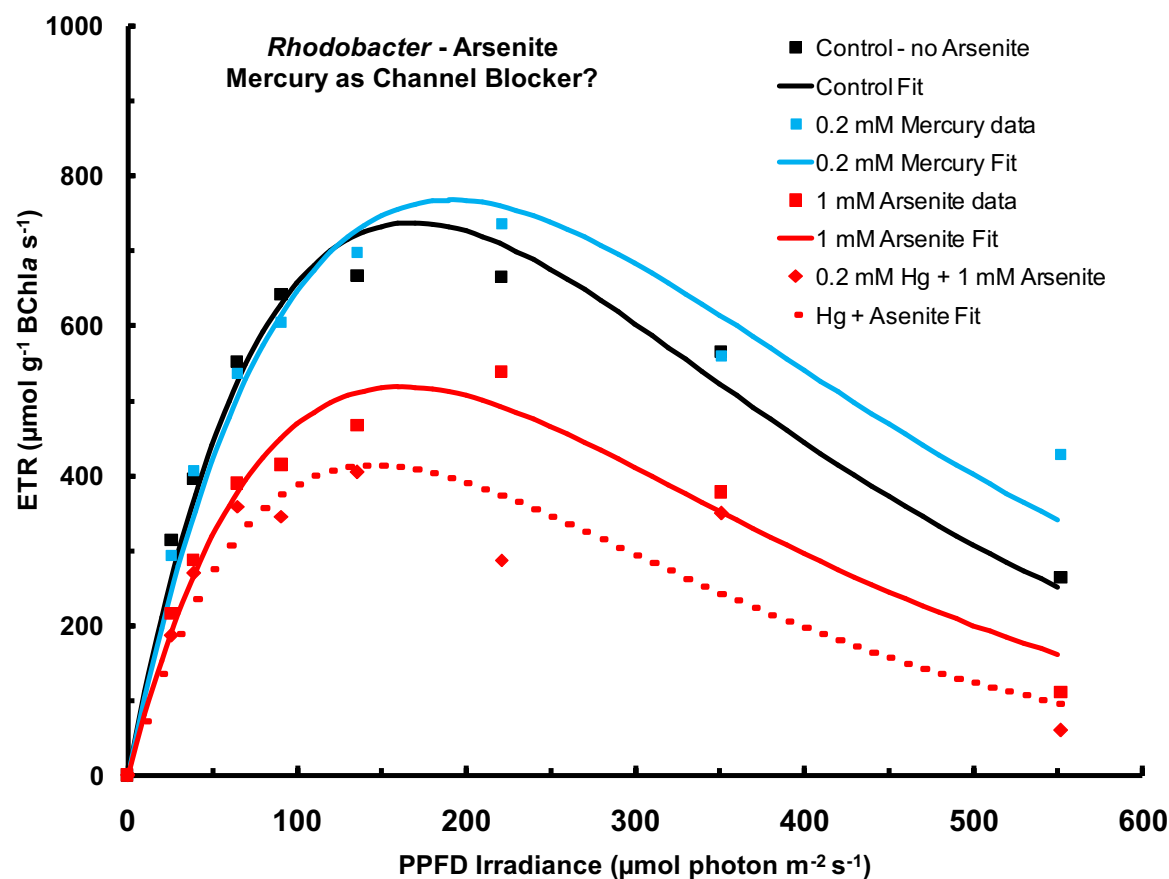


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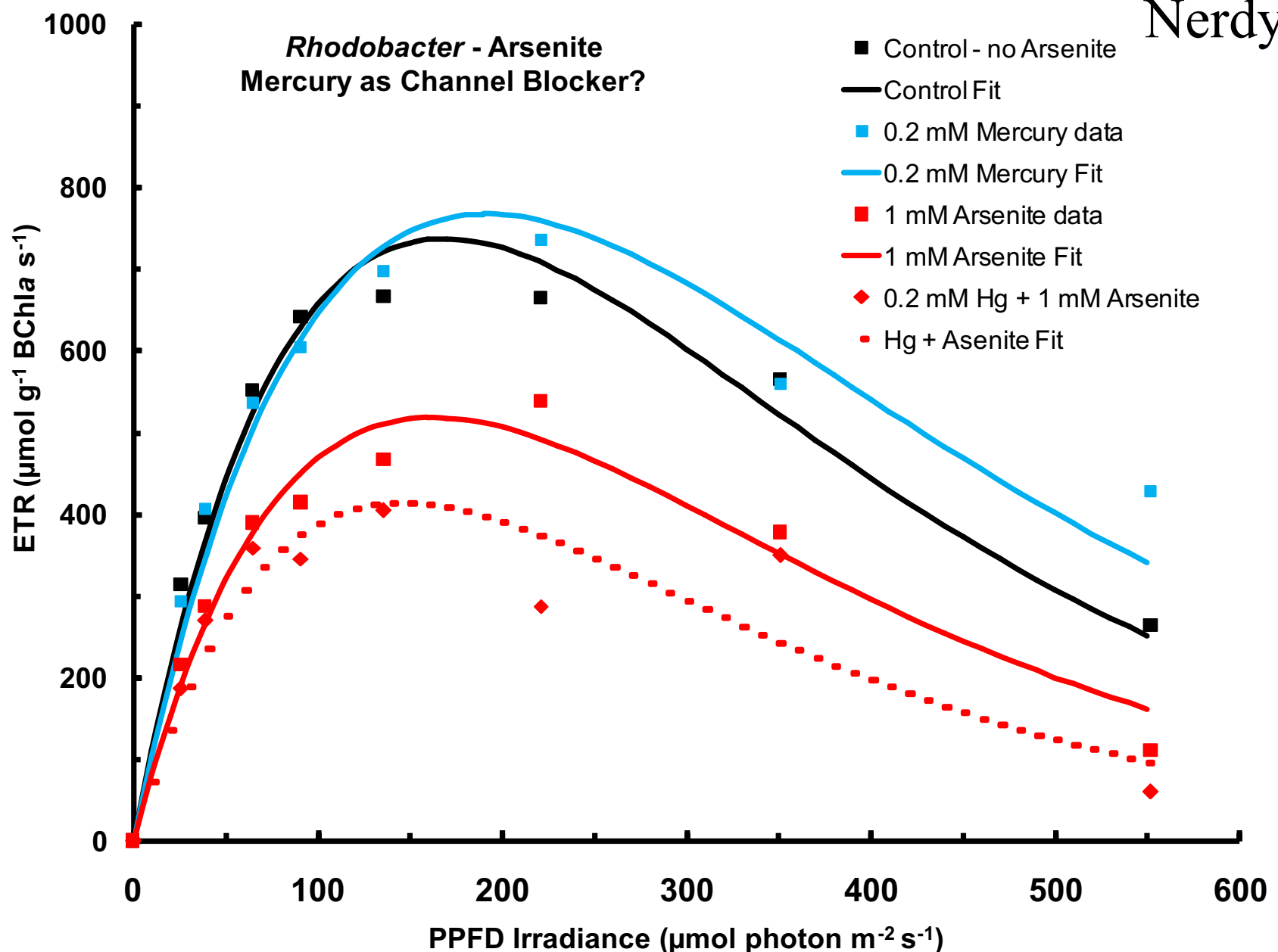


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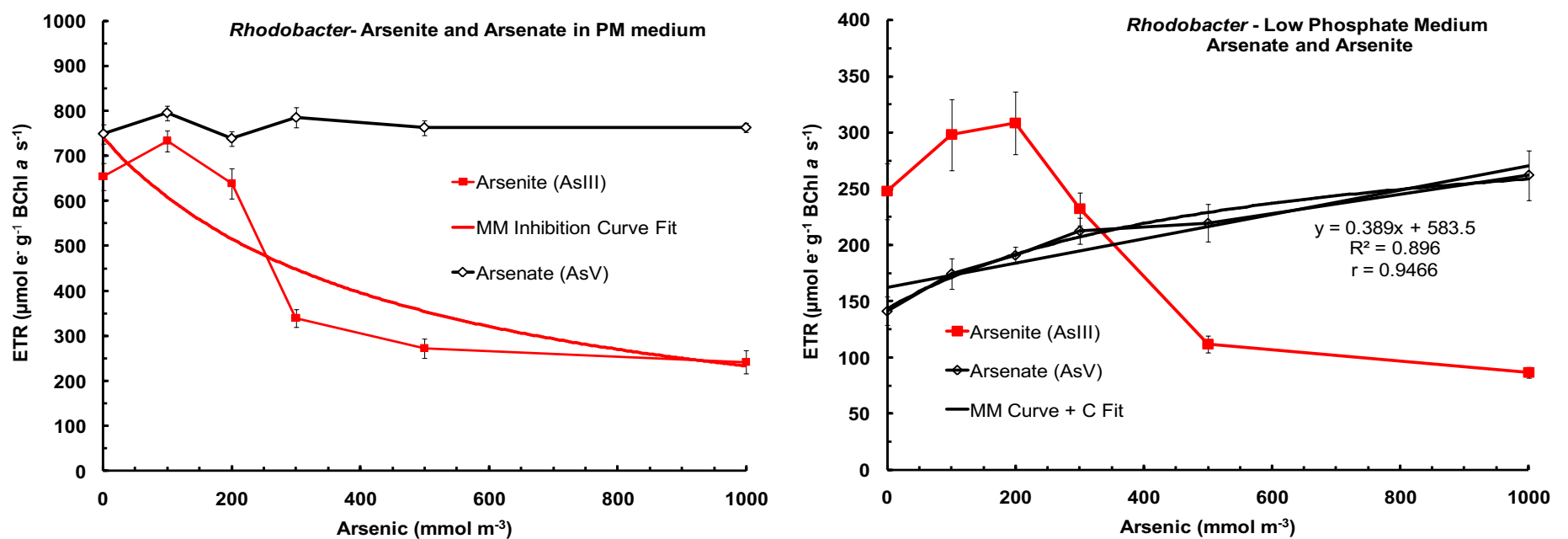


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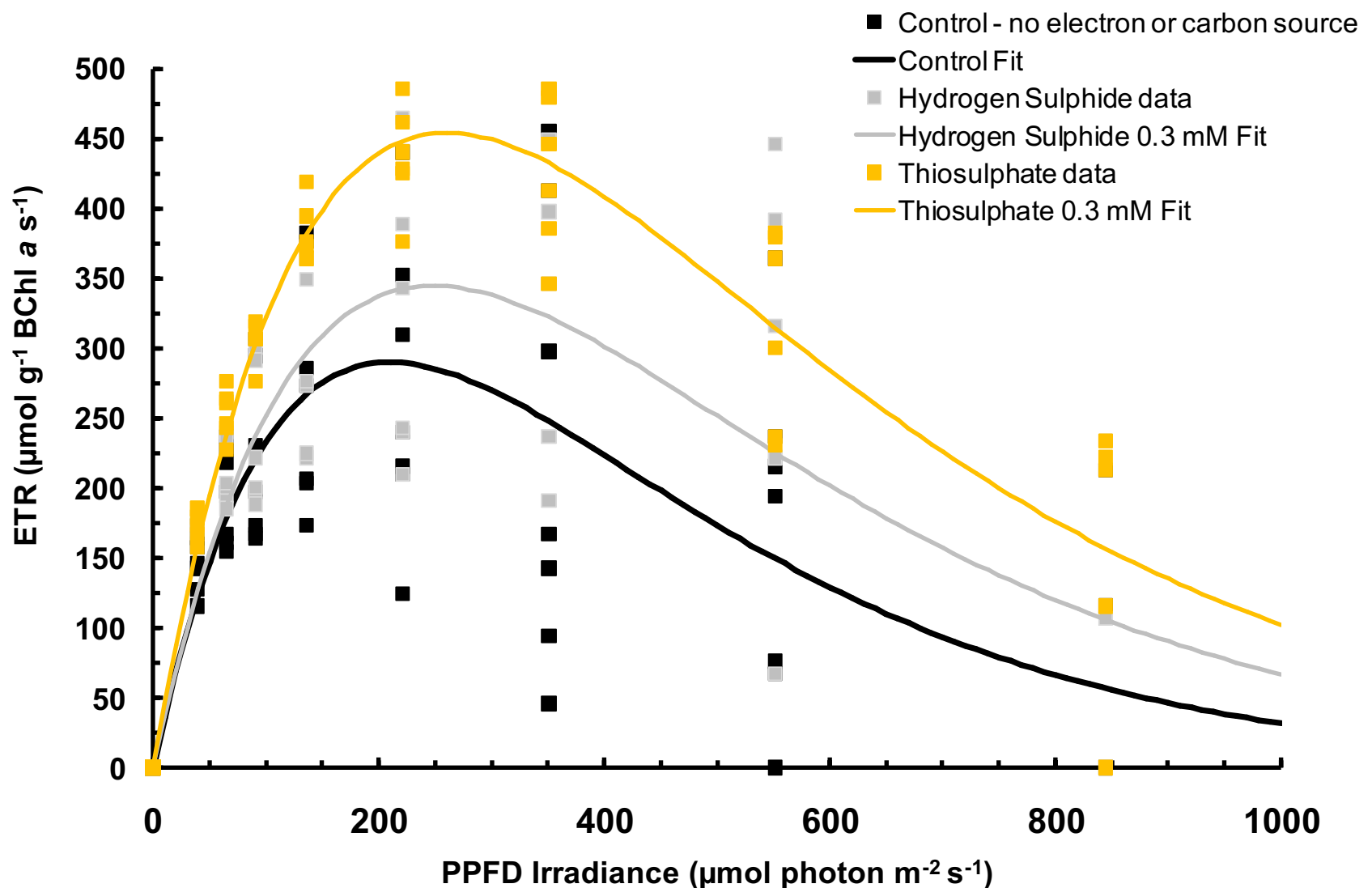
Rhodobacter - H_2S and SO_3^{2-} as e^- sources

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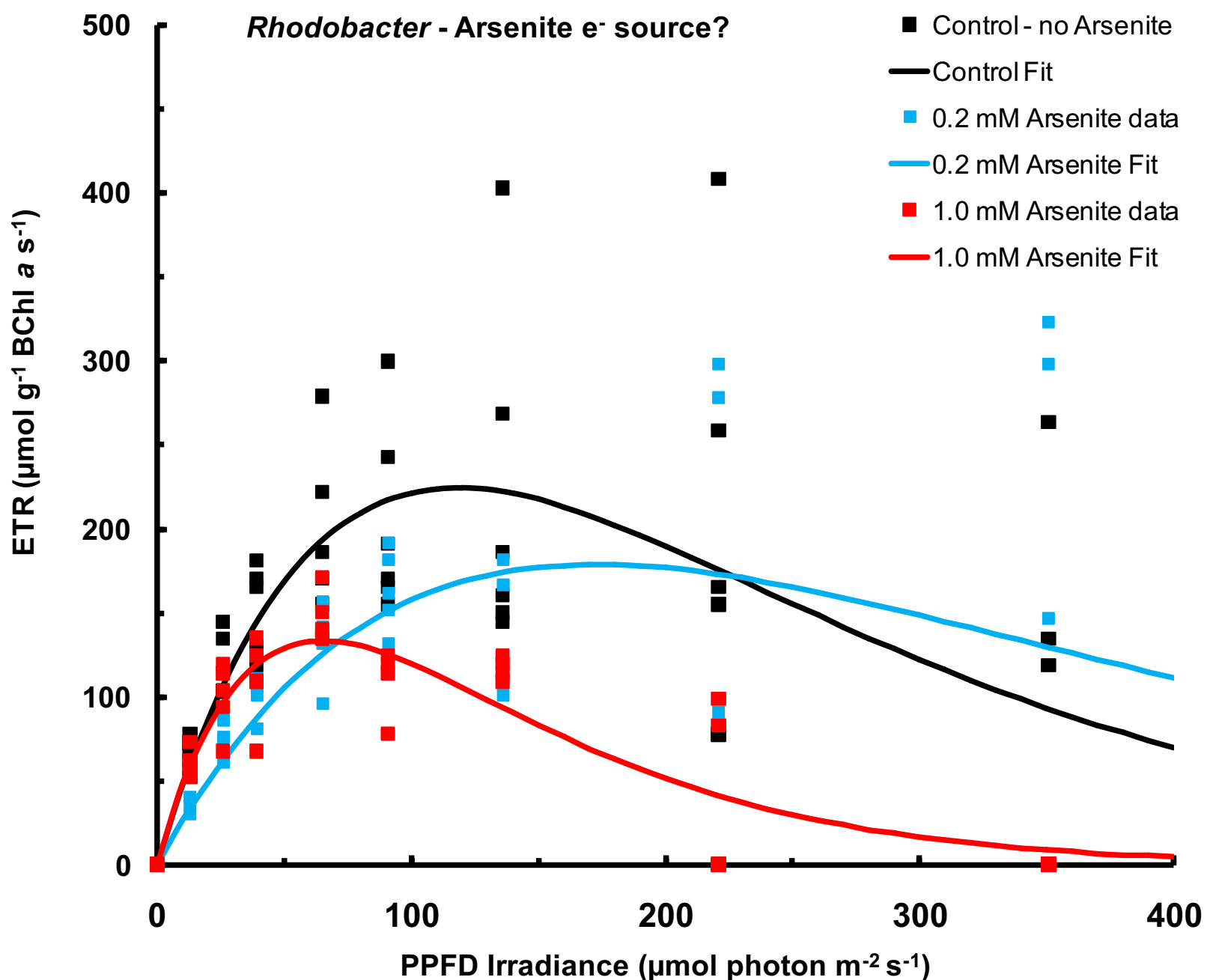


Fig. 6 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with 0.2 mol m⁻³ As(III) and 1 mol m⁻³ As(III). As(III) did not increase the photosynthetic ETR and so As(III) could not be used as an electron source for photosynthesis (cf. Ref 5)

Conclusions

- **Arsenite (As(III)) is toxic but Arsenate (As(V)) has no short-term inhibitory effects. Response to As(V) is different under high P compared to low-P conditions.**
- **Al(OH)₃ and Hg²⁺ are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.**
- **As(III) does not act as an electron source for photosynthesis in *Rhodobacter*.**

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How to Construct the Poster



Arsenic toxicity in a Photosynthetic Bacterial Symbiont of *Wolffia aarhiza*.



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Summary

An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia aarhiza* plants has been cultured and identified as a *Rhodopseudomonad* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a Rhodopseudomonad. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorptance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as mol e⁻ g⁻¹ BChl *a* s⁻¹. ETR vs. Irradiance (E) curves fitted the waiting-in-line model ($ETR = (ETR_{max} \times E/E_{opt}) \times \exp(1 - E/E_{opt})$). Yield (Y) was only ≈ 0.3 to 0.4 . *Rhodobacter* saturates at about 250 to 350 $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$ or $\approx 15\%$ sunlight and shows photoinhibition at high irradiances (overall E_{opt} was $298 \pm 7.36 \mu\text{mol quanta m}^{-2} \text{ s}^{-1}$; $ETR_{max} = 642 \pm 10.6 \mu\text{mol e}^{-} \text{ g}^{-1} \text{ BChl } a \text{ s}^{-1}$; $\alpha = 6.05 \pm 0.200 \text{ e}^{-} \text{ photon}^{-1} \text{ m}^2 \text{ g}^{-1} \text{ BChl } a$). Photosynthetic performance was much worse in Low-P medium ($n = 108$, overall $E_{opt} 158 \pm 15.4 \mu\text{mol quanta m}^{-2} \text{ s}^{-1}$; $ETR_{max} = 194 \pm 13.5 \mu\text{mol e}^{-} \text{ g}^{-1} \text{ BChl } a \text{ s}^{-1}$; $\alpha = 3.30 \pm 0.400 \text{ e}^{-} \text{ photon}^{-1} \text{ m}^2 \text{ g}^{-1} \text{ BChl } a$). *Rhodobacter* is resistant to As (V) toxicity up to at least 1 mol m^{-3} in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was $497 \pm 100 \text{ mmol m}^{-3}$ but there is a threshold effect below about 200 mmol m^{-3} As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. Al(OH)₃ and Hg²⁺ are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

Keywords: Photosynthetic bacteria, *Rhodobacter*, anoxygenic photosynthesis, integrating sphere spectrophotometry, PAM fluorometry.

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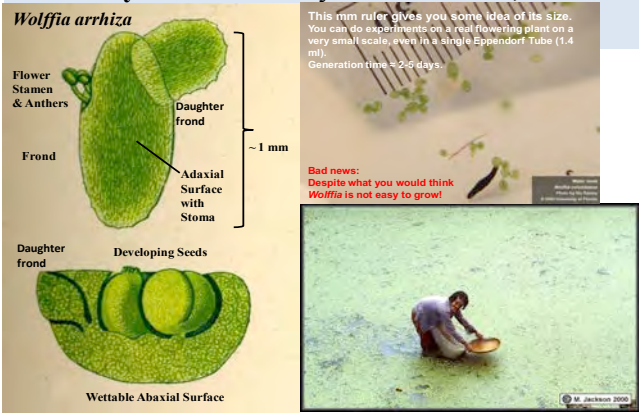
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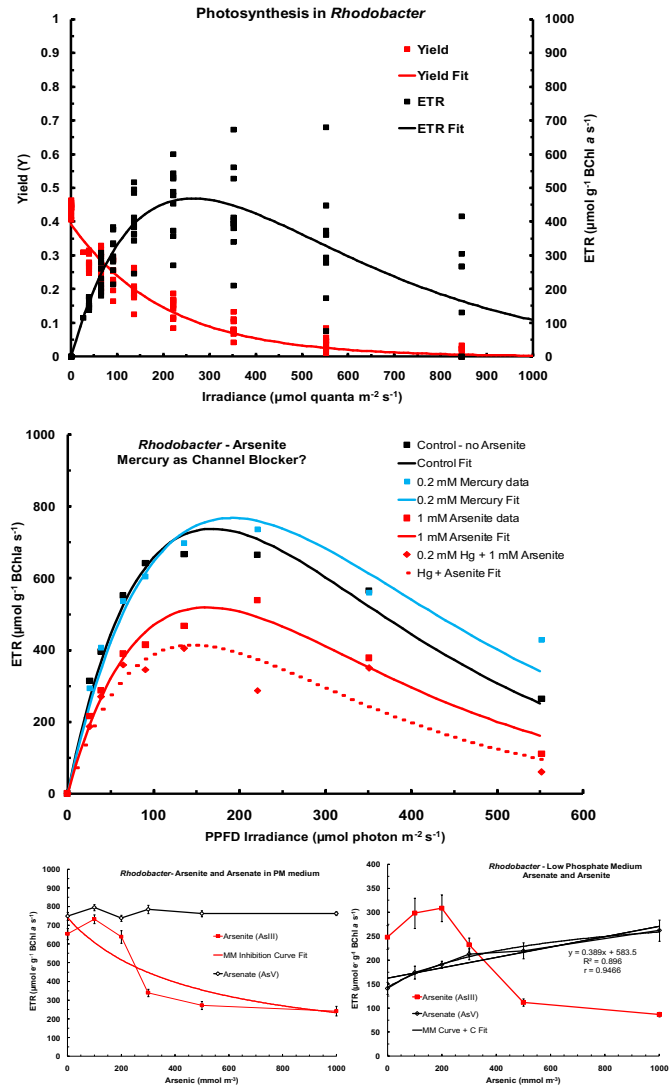
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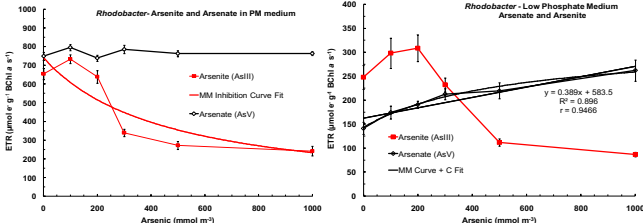
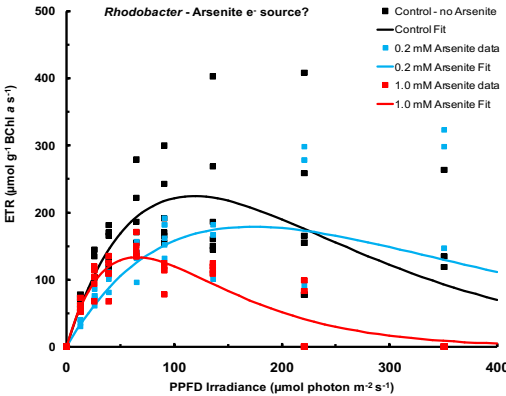
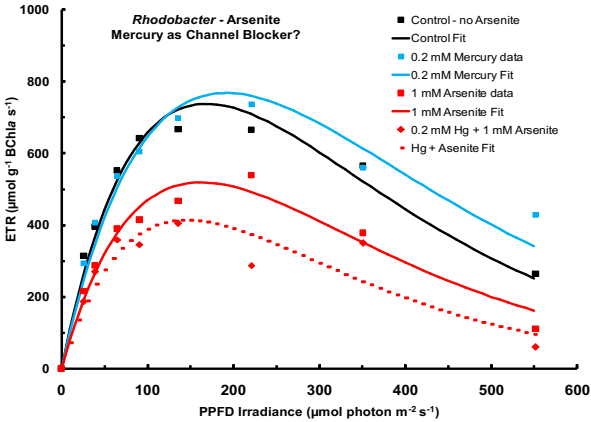
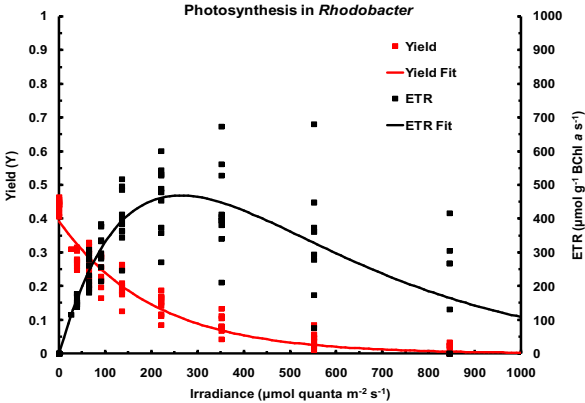
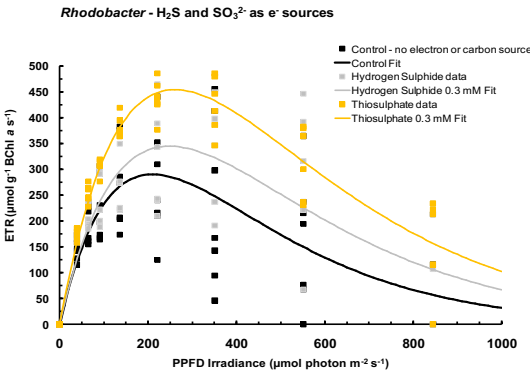
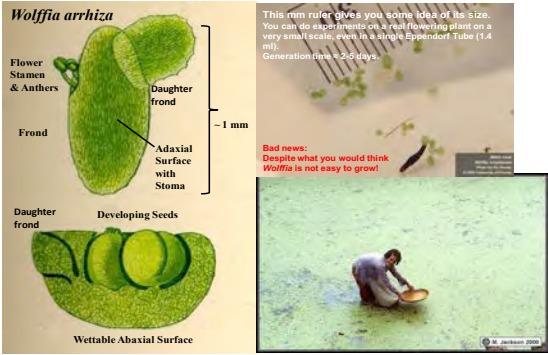
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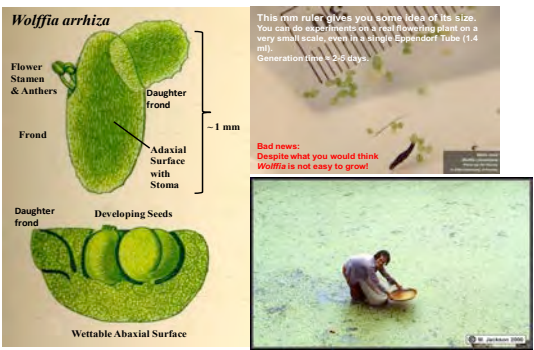


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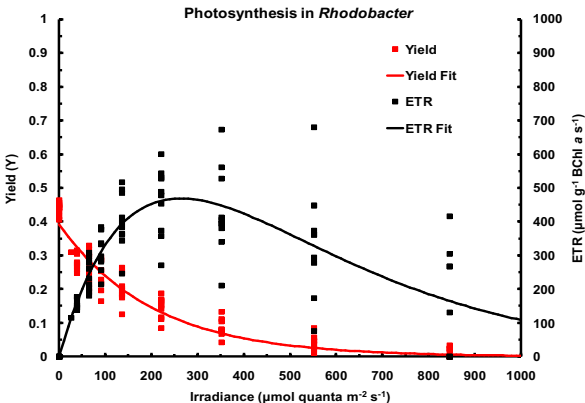
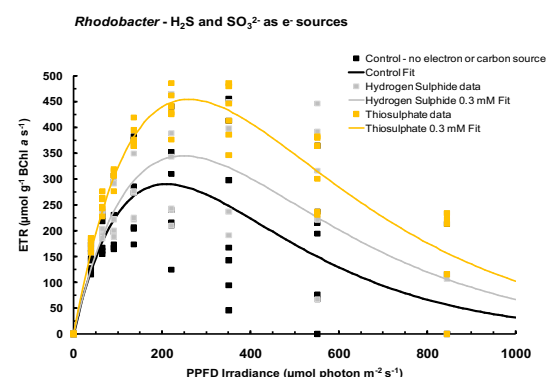


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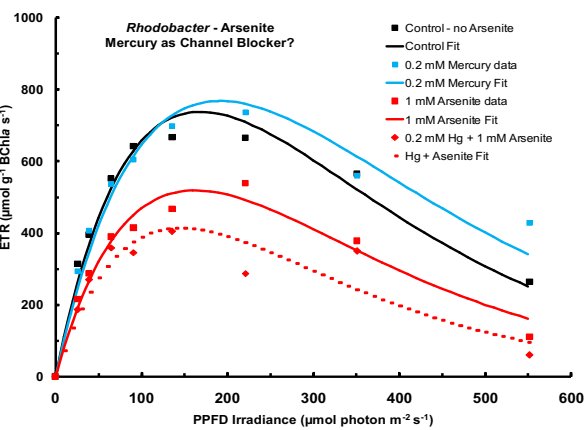
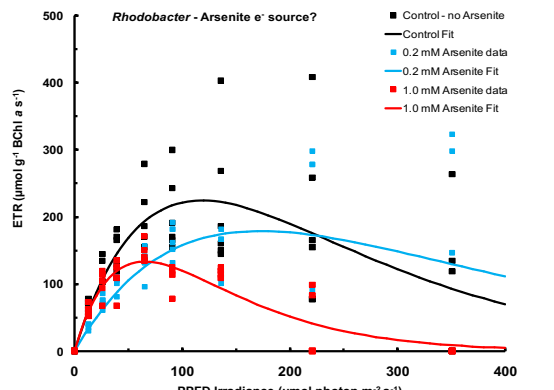


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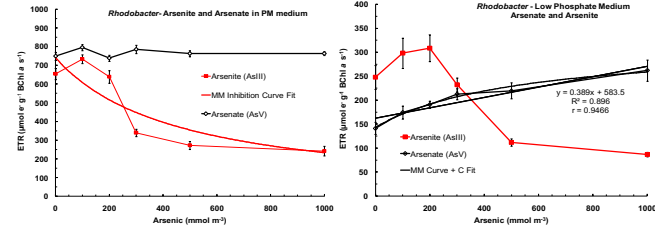


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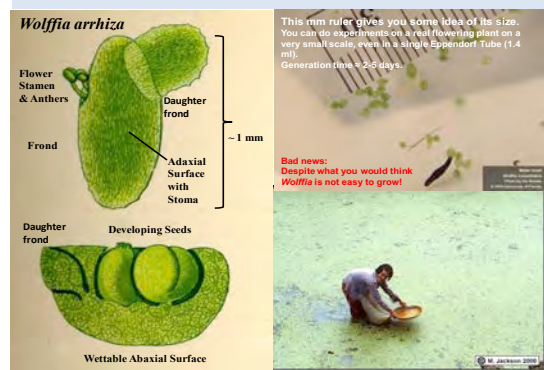


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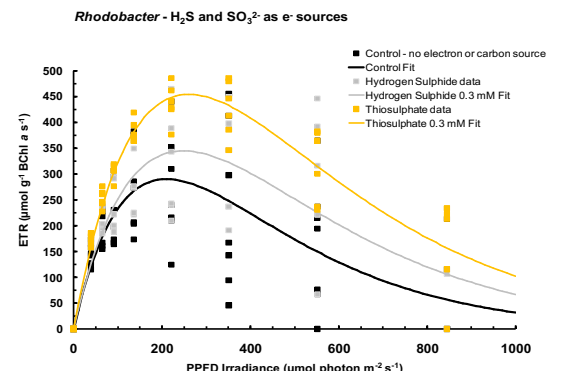
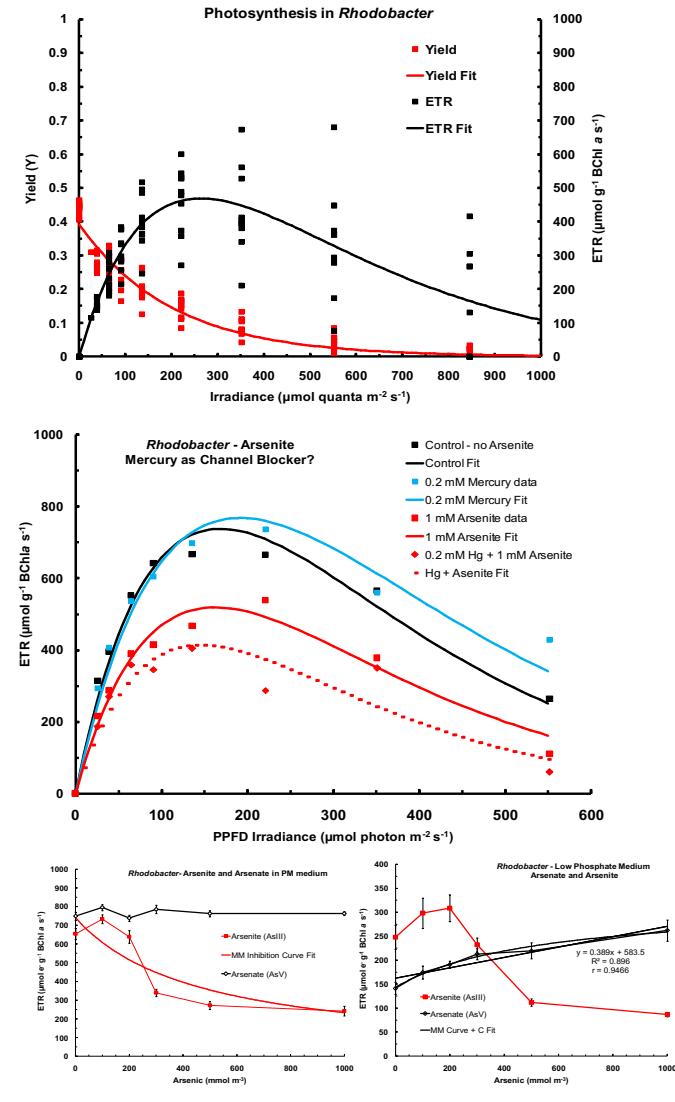


Fig. 5 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with Na₂S or thiosulphate (300 mmol m⁻³). H₂S marginally increased the photosynthetic ETR and so was being used as an electron source for photosynthesis. Thiosulphate stimulated ETR by more than 50% and so was being used as an electron source. Similar experiments using Fe(II) as a potential e⁻ source showed that *Rhodobacter* could not use Fe(II) as an electron source unlike most PS Bacteria (Refs 1,2,3)

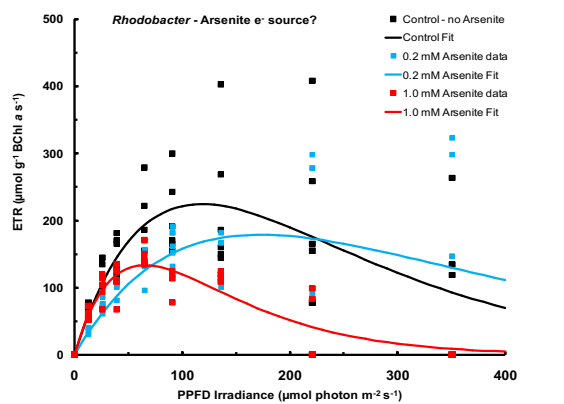


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Summary

An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia arrhiza* plants has been cultured and identified as a *Rhodospseudomonad* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a *Rhodospseudomonad*. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorbance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as $\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$. ETR vs. Irradiance (E) curves fitted the waiting-in-line model ($\text{ETR} = (\text{ETR}_{\text{max}} \times \text{E}/\text{E}_{\text{opt}}) \times \exp(1 - \text{E}/\text{E}_{\text{opt}})$). Yield (Y) was only ≈ 0.3 to 0.4 . *Rhodobacter* saturates at about 250 to $350 \mu\text{mol photons m}^{-2} \text{s}^{-1}$ or $\approx 15\%$ sunlight and shows photoinhibition at high irradiances (overall E_{opt} was $298 \pm 7.36 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 642 \pm 10.6 \mu\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$; $\text{Alpha } \square (\alpha) = 6.05 \pm 0.200 \text{ e}^- \text{ photon}^{-1} \text{ m}^2 \text{ g}^{-1} \text{BChl } a$). Photosynthetic performance was much worse in Low-P medium ($n = 108$, overall $\text{E}_{\text{opt}} 158 \pm 15.4 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 194 \pm 13.5 \mu\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$; $\alpha = 3.30 \pm 0.400 \text{ e}^- \text{ photon}^{-1} \text{ m}^2 \text{ g}^{-1} \text{BChl } a$). *Rhodobacter* is resistant to As (V) toxicity up to at least 1 mol m^{-3} in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was $497 \pm 100 \text{ mmol m}^{-3}$ but there is a threshold effect below about 200 mmol m^{-3} As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

Keywords: Photosynthetic bacteria, *Rhodobacter*, anoxygenic photosynthesis, integrating sphere spectrophotometry, PAM fluorometry.

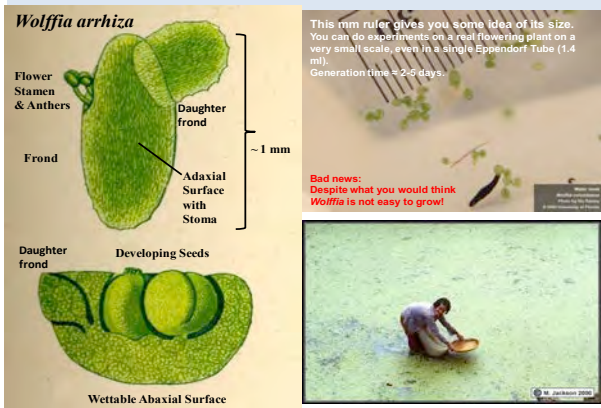


Fig. 1 *Wolffia* is a genus of very simple flowering plants (Angiosperms). They are the smallest known flowering plants. The very simple anatomy of *Wolffia* helps in the analysis and interpretation of arsenic uptake studies. No roots, stems or leaves or xylem and phloem. The very small size is also an advantage in the laboratory.

Wolffia is somewhat resistant to arsenic but is more resistant to Arsenate (AsV) than to Arsenite (AsIII) (ref. 4). *Wolffia* cells are able to interconvert As(III) and As(V) but cannot convert it into volatile methyl arsines. Arsenic accumulation in *Wolffia* is important in Thailand because it is used as a vegetable ("Khai-nam").

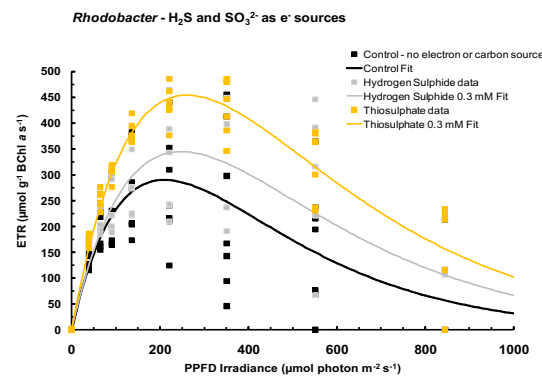


Fig. 2 Like the water fern *Azolla*, *Wolffia* has endosymbiotic bacteria capable of N-fixation. *Wolffia* has an endosymbiotic N-fixing anoxygenic photosynthetic bacterium, *Rhodobacter capsulatus*. It is easily grown in culture in PM medium using acetate as a carbon source using standard methods for photosynthetic bacteria (refs 1,2,3) and its photosynthesis can be easily measured using a Blue-diode PAM on cells filtered onto glass fibre disks (ref 1,2,3). The maximum Yield (Y) of PS bacteria is usually ≈ 0.4 . Rapid light curves can be easily fitted to a Waiting-in-Line curve. Photosynthetic electron transport (ETR) curves were fitted to the Waiting-in-Line Equation ($\text{ETR}_{\text{max}} = 469 \pm 30.1 \mu\text{mol e}^- \text{g}^{-1} \text{BChl } a \text{ s}^{-1}$; $\text{E}_{\text{opt}} = 263 \pm 24.8 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{Alpha } \square (\alpha) = 4.843 \pm 0.553 \text{ e}^- \text{ photon}^{-1} \text{ m}^2 \text{ g}^{-1} \text{BChl } a$, $n = 72$, $r = 0.8345$, $p < 0.001$).

Fig. 3 Photosynthetic electron transport rate of *Rhodobacter* incubated in PM medium (control) and supplied with $200 \text{ mmol m}^{-3} \text{Hg}^{2+}$ as a potential channel blocking agent compared to cells exposed to arsenite (As(III)). Hg^{2+} had no effect on ETR compared to the control. $1.0 \text{ mol m}^{-3} \text{As (III)}$ was inhibitory by about 50% on ETR_{max} but did not appear to have much effect on the optimum irradiance (E_{opt}) about $200 \mu\text{mol photon m}^{-2} \text{s}^{-1}$. Hg^{2+} had no protective effect against As(III) toxicity. Similar experiments using $\text{Al}(\text{OH})_3$ as a potential channel blocker also showed no effective channel blocking behaviour to prevent As(III) toxicity.

Fig. 5 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with Na_2S or thiosulphate (300 mmol m^{-3}). H_2S marginally increased the photosynthetic ETR and so was being used as an electron source for photosynthesis. Thiosulphate stimulated ETR by more than 50% and so was being used as an electron source. Similar experiments using Fe(II) as a potential e^- source showed that *Rhodobacter* could not use Fe(II) as an electron source unlike most PS Bacteria (Refs 1,2,3)

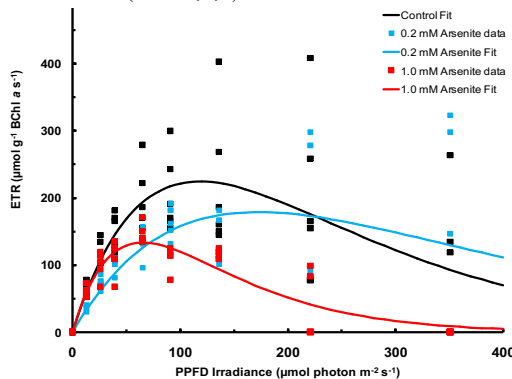
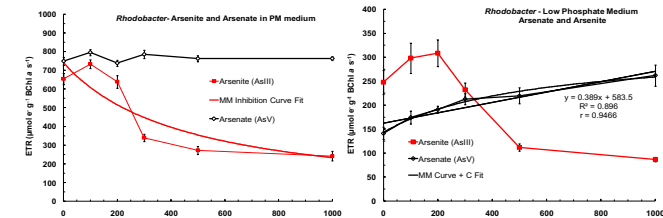
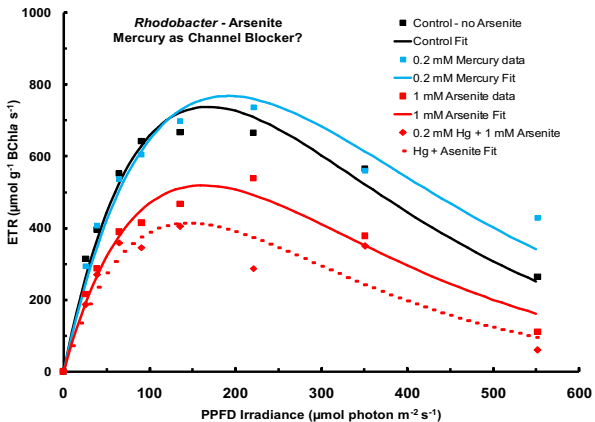
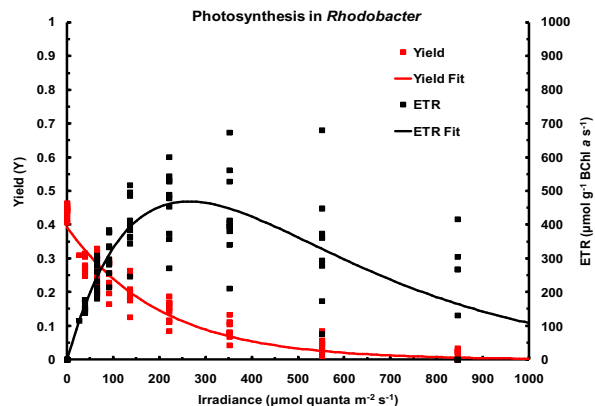


Fig. 6 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with $0.2 \text{ mol m}^{-3} \text{As(III)}$ and $1 \text{ mol m}^{-3} \text{As(III)}$. As(III) did not increase the photosynthetic ETR and so As(III) could not be used as an electron source for photosynthesis (cf. Ref 5)

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References & Wombat.

Conclusions

- Arsenite (As(III)) is toxic but Arsenate (As(V)) has no short-term inhibitory effects. Response to As(V) is different under high P compared to low-P conditions.
- $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.
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References

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Arsenic toxicity in a Photosynthetic Bacterial Symbiont of *Wolffia arrhiza*.



Raymond J. RITCHIE¹ and Siriporn NAKPHET¹

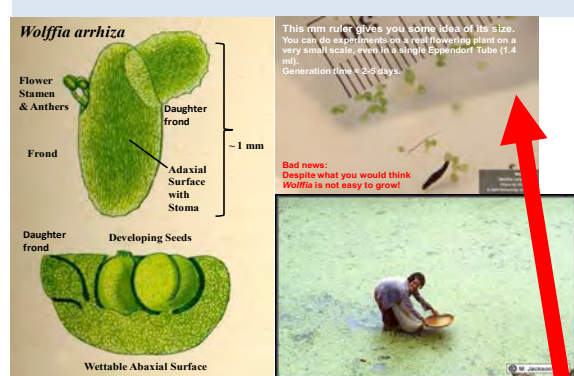
¹Technology and Environment, Prince of Songkla University-Phuket, Kathu, 83120, Thailand,

Corresponding author: Raymond J. RITCHIE, E-mail: raymond.ritchie@uni.sydney.edu.au, raymond.r@phuket.psu.ac.th

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An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia arrhiza* plants has been cultured and identified as a *Rhodopseudomonad* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a Rhodopseudomonad. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorbance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as mol e⁻ g⁻¹ BChl *a* s⁻¹. ETR vs. Irradiance (E) curves fitted the waiting-in-line model (ETR = (ETR_{max} × E/E_{opt}) × exp (1-E/E_{opt})). Yield (Y) was only ≈ 0.3 to 0.4. *Rhodobacter* saturates at about 250 to 350 μmol photons m⁻² s⁻¹ or ≈ 15% sunlight and shows photoinhibition at high irradiances (overall E_{opt} was 298 ± 7.36 μmol quanta m⁻² s⁻¹; ETR_{max} = 642 ± 10.6 μmol e⁻ g⁻¹ BChl *a* s⁻¹; Alpha (α) = 6.05 ± 0.200 e⁻ photon⁻¹ m² g⁻¹ BChl *a*). Photosynthetic performance was much worse in Low-P medium (n = 108, overall E_{opt} 158± 15.4 μmol quanta m⁻² s⁻¹; ETR_{max} = 194 ± 13.5 μmol e⁻ g⁻¹ BChl *a* s⁻¹; α = 3.30 ± 0.400 e⁻ photon⁻¹ m² g⁻¹ BChl *a*). *Rhodobacter* is resistant to As (V) toxicity up to at least 1 mol m⁻³ in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was 497 ± 100 mmol m⁻³ but there is a threshold effect below about 200 mmol m⁻³ As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. Al(OH)₃ and Hg²⁺ are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

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Final Steps

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• **Check what it looks like at full screen (A1 Size).**

• **Can you read everything from 1.5 m away?**

• **Any post-script file problems?**

• **Make sure you use a postscript font.**

• **Group the entire poster.**

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• **Make a pdf of the grouped poster and get the pdf printed..**

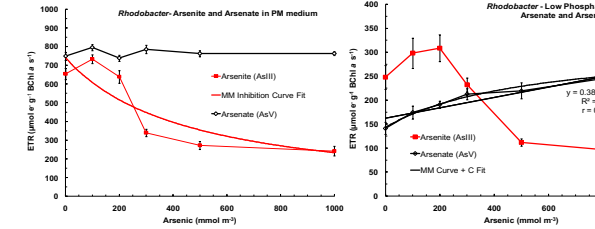
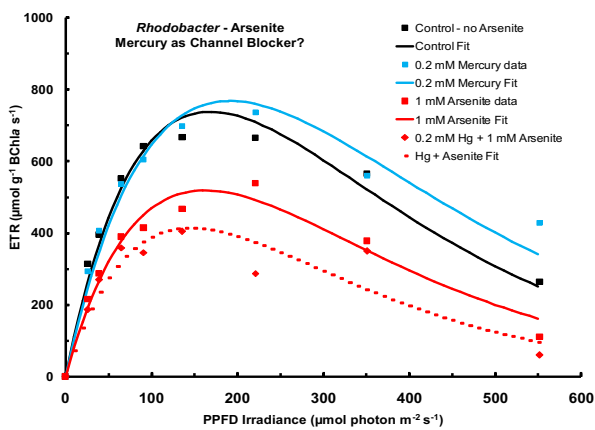
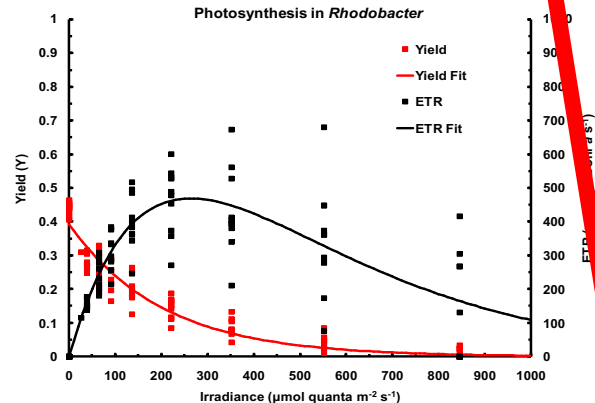
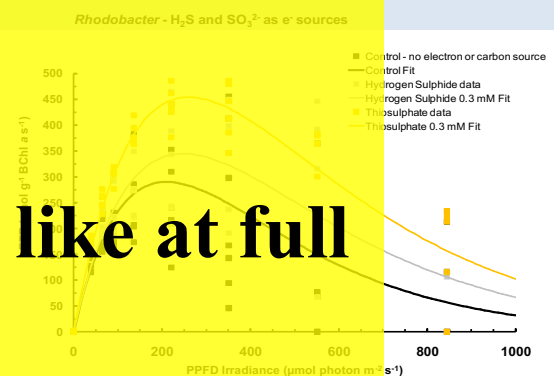


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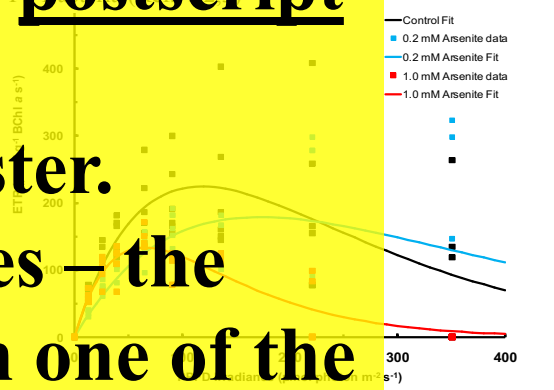


Fig. 6 Photosynthetic electron transport rate of *Rhodopseudomonas* incubated in electron source-free PM medium (control) and supplied with 0.2 mol m⁻³ As(III) and 1 mol m⁻³ As(III). As(III) did not stimulate ETR and so As(III) could not be used as an electron source for photosynthesis

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Some things wrong with the draft poster.

This mm ruler gives you some idea of its size. You can do experiments on a real flowering plant on a very small scale, even in a single Eppendorf Tube (1.4 ml).
Generation time \approx 2-5 days.

White print never works very well. Avoid.
Red print is almost as bad.

Bad news:
Despite what you would think
Wolffia is not easy to grow!

Water meal
Wolffia columbiana
Photo by Vic Ramey
© 2000 University of Florida

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Wolffia columbiana
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Arsenic toxicity in a Photosynthetic Bacterial Symbiont of *Wolffia arrhiza*.



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Final Steps

• Fix text

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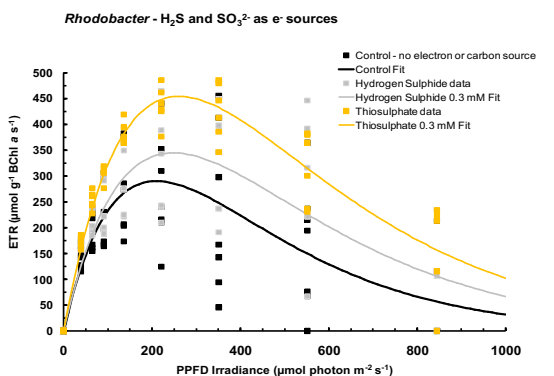


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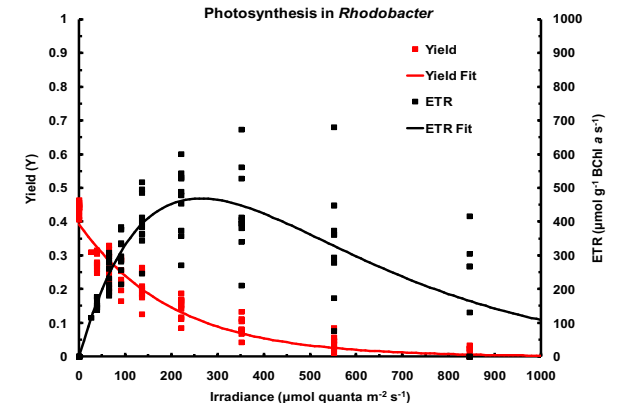


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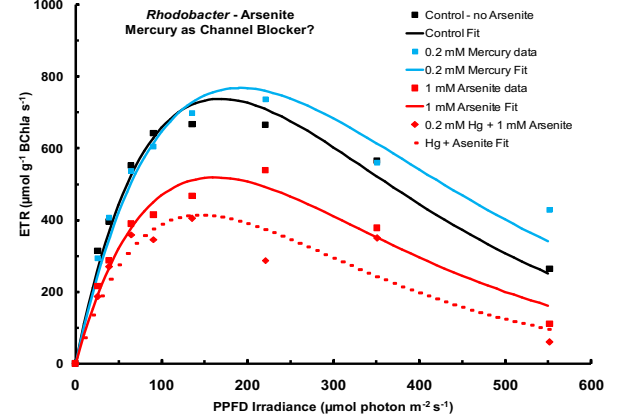


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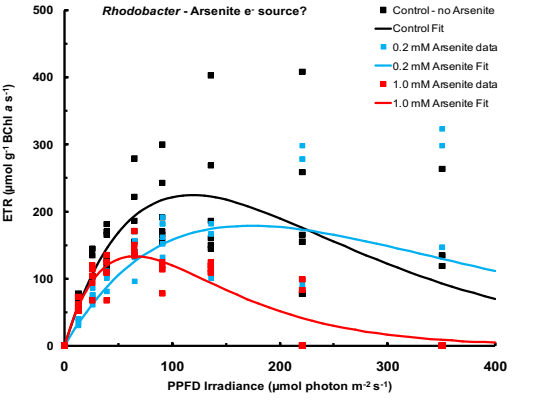


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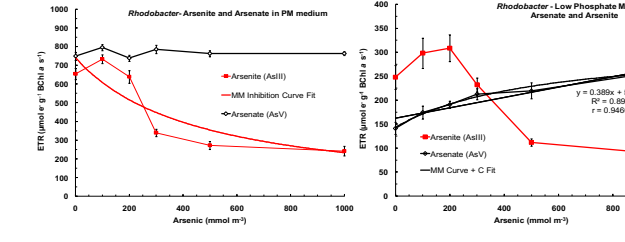


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Summary

An arsenic resistant nitrogen fixing photosynthetic bacterium found to live inside *Wolffia aarhiza* plants has been cultured and identified as a *Rhodospseudomonad* species, most likely a strain of *Rhodobacter capsulatus*. It has BChl *a* as its primary photosynthetic pigment and has spectral properties typical of a *Rhodospseudomonad*. Blue-diode-based PAM (Pulse Amplitude Modulation) technology can be used to measure the photosynthetic electron transport rate (ETR) of the organism. The absorbance of the *Rhodobacter* films on glass fibre discs was measured and used to calculate actual ETR as $\mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$. ETR vs. Irradiance (E) curves fitted the waiting-in-line model ($\text{ETR} = (\text{ETR}_{\text{max}} \times \text{E}/\text{E}_{\text{opt}}) \times \exp(1 - \text{E}/\text{E}_{\text{opt}})$). Yield (Y) was only ≈ 0.3 to 0.4 . *Rhodobacter* saturates at about 250 to 350 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ or $\approx 15\%$ sunlight and shows photoinhibition at high irradiances (overall E_{opt} was $298 \pm 7.36 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 642 \pm 10.6 \mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$; Alpha (α) = $6.05 \pm 0.200 \text{ e}^- \text{photon}^{-1} \text{m}^2 \text{g}^{-1} \text{BChl a}$). Photosynthetic performance was much worse in Low-P medium ($n = 108$, overall $\text{E}_{\text{opt}} 158 \pm 15.4 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; $\text{ETR}_{\text{max}} = 194 \pm 13.5 \mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$; $\alpha = 3.30 \pm 0.400 \text{ e}^- \text{photon}^{-1} \text{m}^2 \text{g}^{-1} \text{BChl a}$). *Rhodobacter* is resistant to As (V) toxicity up to at least 1 mol m^{-3} in high and low P medium. The K_i for As(III) in High and Low-P are not significantly different: overall mean was $497 \pm 100 \text{ mmol m}^{-3}$ but there is a threshold effect below about 200 mmol m^{-3} As(III). Fe(II) and As(III) did not appear to act as electron sources but thiosulphate did act as an electron source for photosynthesis. $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.

Keywords: Photosynthetic bacteria, *Rhodobacter*, anoxygenic photosynthesis, integrating sphere spectrophotometry, PAM fluorometry.



Fig. 1 *Wolffia* is a genus of very simple flowering plants (Angiosperms). They are the smallest known flowering plants. The very simple anatomy of *Wolffia* helps in the analysis and interpretation of arsenic uptake studies. No roots, stems or leaves or xylem and phloem. The very small size is also an advantage in the laboratory.

Wolffia is somewhat resistant to arsenic but is more resistant to Arsenate (AsV) than to Arsenite (AsIII) (ref. 4). *Wolffia* cells are able to interconvert As(III) and As(V) but cannot convert it into volatile methyl arsines.

Arsenic accumulation in *Wolffia* is important in Thailand because it is used as a vegetable ("Khai-nam").

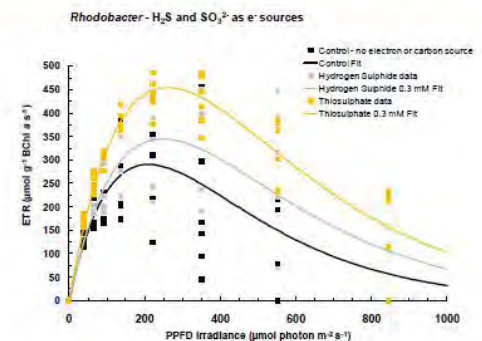


Fig. 2 Like the water fern *Azolla*, *Wolffia* has endosymbiotic bacteria capable of N-fixation. *Wolffia* has an endosymbiotic N-fixing anoxygenic photosynthetic bacterium, *Rhodobacter capsulatus*. It is easily grown in culture in PM medium using acetate as a carbon source using standard methods for photosynthetic bacteria (refs 1,2,3) and its photosynthesis can be easily measured using a Blue-diode PAM on cells filtered onto glass fibre disks (ref 1,2,3). The maximum Yield (Y) of PS bacteria is usually ≈ 0.4 . Rapid light curves can be easily fitted to a Waiting-in-Line curve. Photosynthetic electron transport (ETR) curves were fitted to the Waiting-in-Line Equation ($\text{ETR}_{\text{max}} = 469 \pm 30.1 \mu\text{mol e}^- \text{g}^{-1} \text{BChl a s}^{-1}$; $\text{E}_{\text{opt}} = 263 \pm 24.8 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$; Alpha (α) = $4.843 \pm 0.553 \text{ e}^- \text{photon}^{-1} \text{m}^2 \text{g}^{-1} \text{BChl a}$, $n = 72$, $r = 0.8345$, $p < 0.001$).

Fig. 5 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with Na_2S or thiosulphate (300 mmol m^{-3}). H_2S marginally increased the photosynthetic ETR and so was being used as an electron source for photosynthesis. Thiosulphate stimulated ETR by more than 50% and so was being used as an electron source. Similar experiments using Fe(II) as a potential e^- source showed that *Rhodobacter* could not use Fe(II) as an electron source unlike most PS Bacteria (Refs 1,2,3)

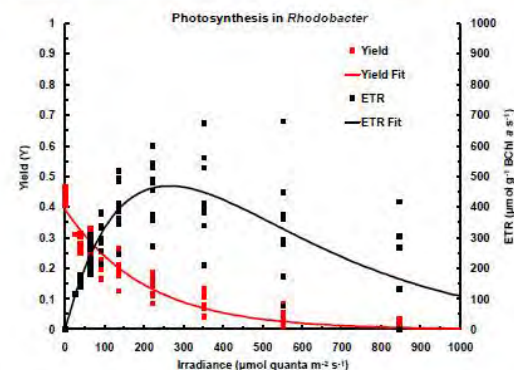


Fig. 3 Photosynthetic electron transport rate of *Rhodobacter* incubated in PM medium (control) and supplied with $200 \text{ mmol m}^{-3} \text{Hg}^{2+}$ as a potential channel blocking agent compared to cells exposed to arsenite (As(III)). Hg^{2+} had no effect on ETR compared to the control. $1.0 \text{ mol m}^{-3} \text{As(III)}$ was inhibitory by about 50% on ETR_{max} but did not appear to have much effect on the optimum irradiance (E_{opt}) about $200 \mu\text{mol photon m}^{-2} \text{s}^{-1}$. Hg^{2+} had no protective effect against As(III) toxicity. Similar experiments using $\text{Al}(\text{OH})_3$ as a potential channel blocker also showed no effective channel blocking behaviour to prevent As(III) toxicity.

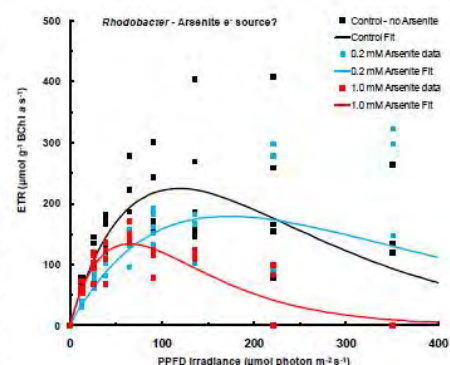


Fig. 6 Photosynthetic electron transport rate of *Rhodobacter* incubated in electron source-free PM medium (control) and supplied with $0.2 \text{ mol m}^{-3} \text{As(III)}$ and $1 \text{ mol m}^{-3} \text{As(III)}$. As(III) did not increase the photosynthetic ETR and so As(III) could not be used as an electron source for photosynthesis (cf. Ref 5)

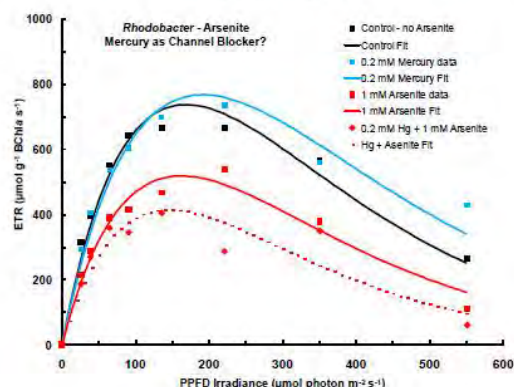


Fig. 4 The toxicity of As(III) and As(V) are quite different after 3h exposure to arsenic. As(III) is obviously toxic and shows approximately Michaelis-Menten kinetics ($K_i \approx 0.5 \text{ mmol m}^{-3}$) but with a threshold effect and is unaffected by [Phosphate]. *Rhodobacter* is highly resistant to As(V) and shows no toxicity in cells grown in PM medium which contains very high Pi ($\approx 10 \text{ mmol m}^{-3} \text{P}$); in low P (0.1 mmol m^{-3}) there is significant stimulation of photosynthesis.

Conclusions

- Arsenite (As(III)) is toxic but Arsenate (As(V)) has no short-term inhibitory effects. Response to As(V) is different under high P compared to low-P conditions.
- $\text{Al}(\text{OH})_3$ and Hg^{2+} are common aquaporin channel blocking agents but neither acted as protectants against As(III) toxicity.
- As(III) does not act as an electron source for photosynthesis in *Rhodobacter*.

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